A Matter of Fact



Mixtures and Pure Substances

- Scientists like to classify things.
- One way that scientists classify matter is by its composition.
- Ultimately, all matter can be classified as mixtures, elements and compounds.

- Mixtures two or more substances that are not chemically combined with each other and can be separated by physical means.
- Solutions a special kind of mixture where one substance dissolves in another.

Homogeneous Mixtures

- A mixture that appears to be the same throughout.
- It is "well mixed."
- The particles that make up the mixture are very small and not easily recognizable.

Examples of homogeneous mixtures



Milk, toothpaste, and mayonnaise are homogeneous mixtures. They are also colloids.

Colloids

- In a colloid the particles are mixed together but not dissolved.
- The particles are relatively large and are kept permanently suspended.



Colloids



- A colloid will not separate upon standing.
- The particles are constantly colliding, and this allows a colloid to scatter light - thus colloids often seem cloudy.

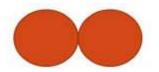
Pure Substances

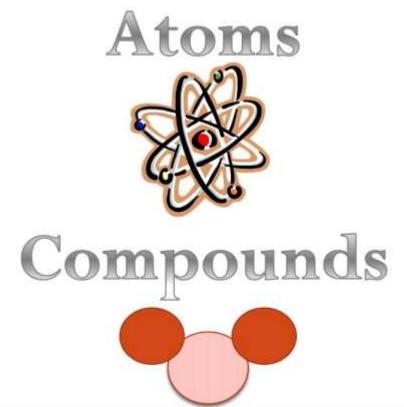
 A sample of matter that has definite chemical and physical properties.

Elements



Molecules





Elements

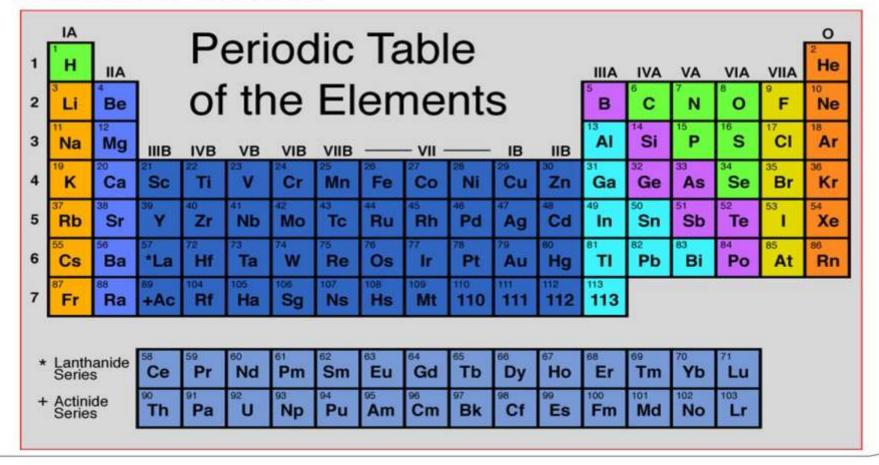
- simplest form of pure substance
- cannot be broken into anything else by physical or chemical means

Compounds

- union of two or more elements
- can be broken into simpler substances by chemical means

Elements

 pure substance that cannot be separated into simpler substance by physical or chemical means.



Compounds

Pure substance composed of two or more different elements joined by chemical bonds.

- Made of elements in a specific ratio that is always the same
- Has a chemical formula
- Can only be separated by chemical means, not physically

CO2

Can you identify the following?

You will be shown a series of photos. Tell if each photo represents an item composed of an element, compound, or mixture.

Review:

- An element contains just one type of atom.
- A compound contains two or more different atoms joined together.
- A mixture contains two or more different substances that are only physically joined together, not chemically.
 - A mixture can contain both elements and compounds.

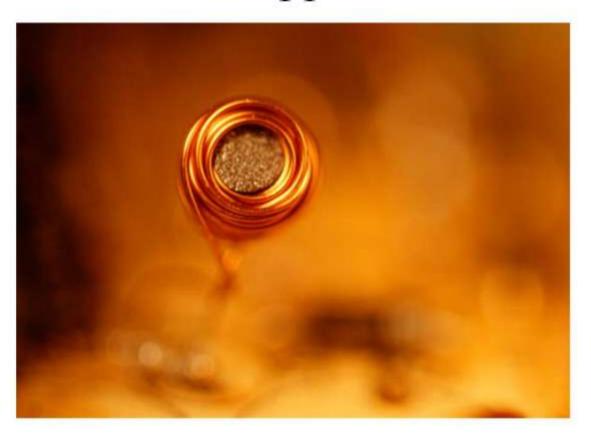






Copper





Element, Compound, or Mixture? Jelly Beans



Element, Compound, or Mixture? Jelly Beans



Element, Compound, or Mixture? Table Sugar

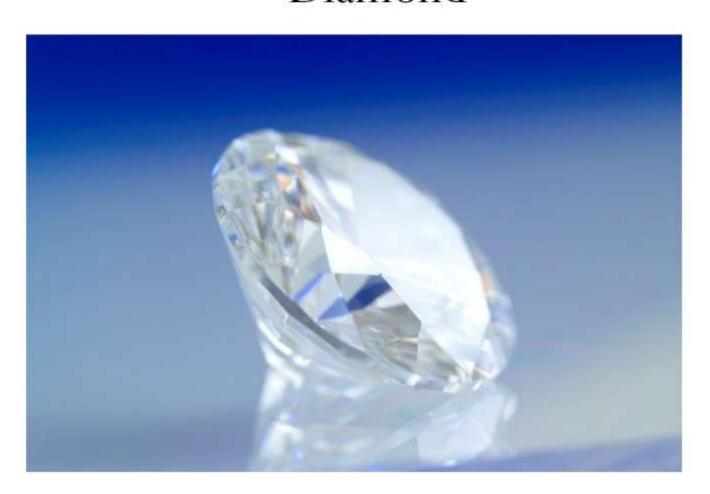


Table Sugar

C12H22O11

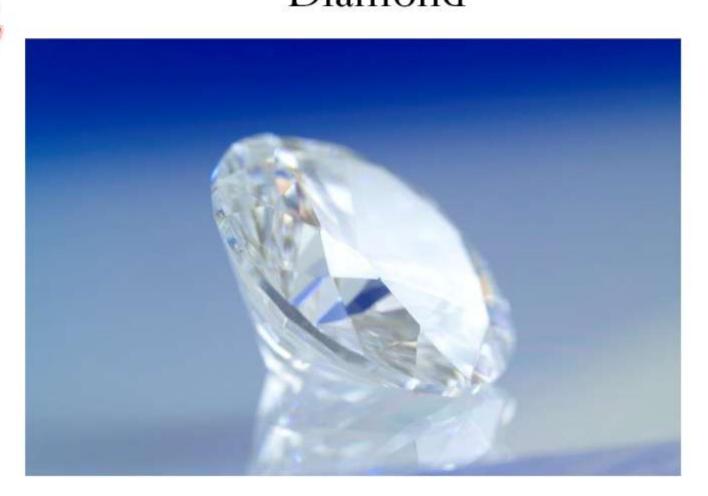


Element, Compound, or Mixture? Diamond



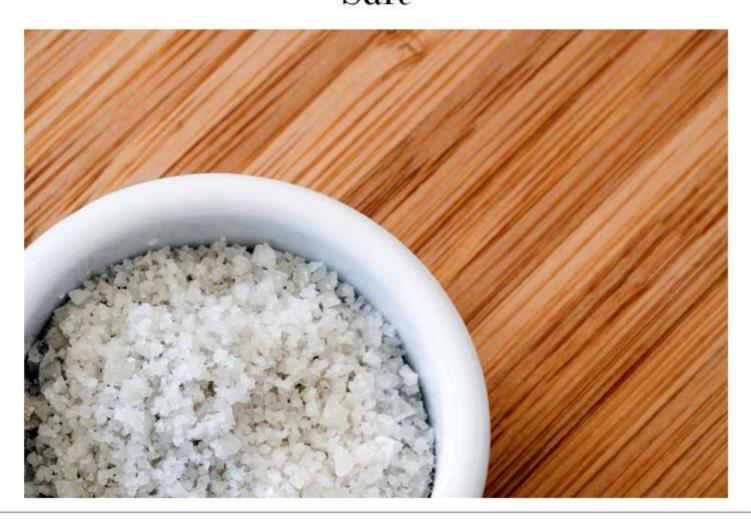
Element, Compound, or Mixture? Diamond

C









Element, Compound, or Mixture? NaCl Salt

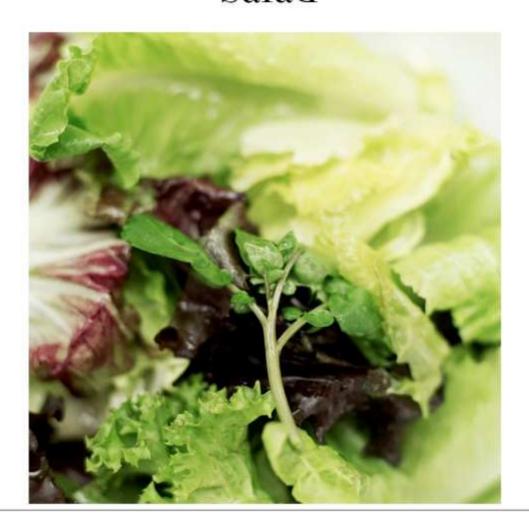


Neon Gas



Neon Gas







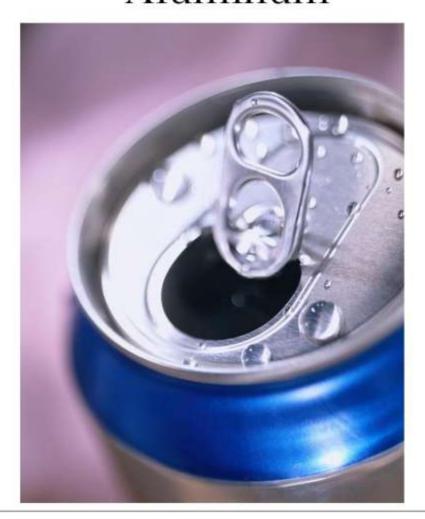
Pure Water



Pure Water

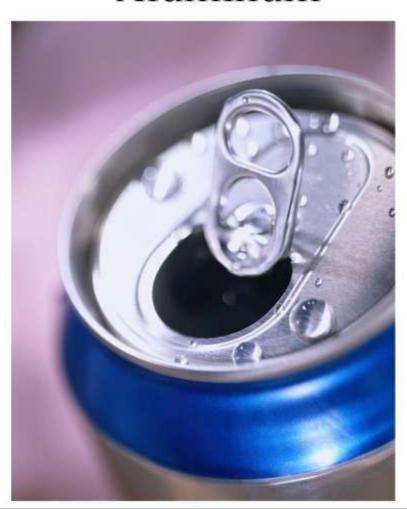


Element, Compound, or Mixture? Aluminum



Aluminum





Lemonade



Lemonade









Sand



Sand



